

1 **Historical trend of polycyclic aromatic hydrocarbons in a sediment core from Osaka**
2 **Bay during the Meghalayan**

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19

20 **Abstract**

21 Polycyclic aromatic hydrocarbons (PAHs) are produced by incomplete combustion of
22 biomass and fossil fuel, yet PAHs have been rarely analyzed in coastal sediment cores as a
23 tracer for human activities before industrialization. The aim of this study was to assess if the
24 historical trend of PAHs can be related to past human activities. To this end, we have

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25 determined the concentrations of PAHs in a 9 m-long sediment core from Osaka Bay, which
26 records history of the last 2400 years. The concentration of PAHs before the beginning of the
27 17th century CE, the beginning of the peaceful Edo period, was consistently low ($< 100 \text{ ng g}^{-1}$)
28 and mainly comprised of smoke-derived PAHs reflecting the natural background. A relative
29 higher abundance of 4–6 ring PAHs from the early 17th century CE and a higher PAH
30 concentration from the early 18th century CE until approximately 1800 CE agreed with a
31 population increase, Cu smelting activities and increasing combustion of charcoal. The
32 constant PAH concentration until the late 19th century CE overlapped with a decline in the
33 population in the Osaka area. An increasing PAH concentration from the late 19th century CE
34 marked the beginning of industrialization in the Modern age. The peak in PAH concentration
35 in 1945 CE was likely caused by burning of wooden structures due to air raids on Osaka City.
36 A second peak around 1980 CE indicated the introduction of cleaner energies. We conclude
37 that PAHs in coastal sediment cores can be used to reconstruct past human activities.

38

39 **Keywords**

40 Osaka Bay, sediment core, Meghalayan, polycyclic aromatic hydrocarbon, industrialization

41

42 **Introduction**

43 For millennia, humans have populated the land around estuaries due to the availability
44 of freshwater, seafood, and fertile land (Allen, 2000; Lotze et al., 2006; Reeder-Myers et al.,
45 2022). Estuaries are prone to accumulate pollutants from various natural and anthropogenic
46 points and diffusive sources originating from their catchments (Barletta et al., 2019; de Souza
47 Machado et al., 2016; Islam and Tanaka, 2004). Sediment cores from coastal areas are
48 valuable archives for reconstructing past climate conditions including sea-surface
49 temperatures (SST), ancient population changes and shifts in social systems (e.g., Kajita et al.,
50 2018, 2023; Kawahata et al., 2017) and early agricultural and smelting activities (Sun et al.,
51 2016; Zong et al., 2010). Furthermore, polycyclic aromatic hydrocarbons (PAHs), a proxy for
52 climate-driven or human-related historical fires and biomass and charcoal burning,
53 accumulate in coastal sediment cores.

54 Humans have burnt biomass and charcoal for millennia through domestic fires, forest
55 clearings, stubble burning, and fueling manufactures such as smelting. More
56 contemporaneous activities are the burning coal, oil, gas for industrial activities and the
57 combustion of gasoline in vehicular traffic. Furthermore, natural wildfires are common
58 around the globe. PAHs are formed as a natural byproduct of incomplete combustion of
59 biomass and carbonaceous fuels (Wilcke, 2007). The carbon and hydrogen content of PAHs
60 partly reflect their source. Diagnostic ratios can be used to distinguish between petroleum,
61 biomass, and coal (Tobiszewski and Namieśnik, 2012; Yunker et al., 2002), which allow us to
62 distinguish plant communities (gymnosperms vs. angiosperms) or smoke and combustion
63 residues (Karp et al., 2020). PAHs enter the marine environment via atmospheric deposition,
64 or directly via oil spills, ship traffic, and river discharge (Baek et al., 1991; Wang et al., 1999;
65 Yunker et al., 2002). PAHs are eventually deposited in the sediment. Owing to their low
66 biodegradability due to the lack of functional groups, the historical trend of PAHs is well
67 preserved in sediment cores from freshwater lakes and ponds. Therefore, PAHs can be related

68 to past human activities such as forest fires, biomass and fossil fuel burning (Argiriadis et al.,
69 2018; Elmquist et al., 2007; Slater et al., 2013; Topness et al., 2023; Vachula et al., 2022; Yan
70 et al., 2005). Furthermore, several studies have discussed the temporal variation in PAHs in
71 coastal sediment cores with reference to industrial and traffic-related emissions (Guo et al.,
72 2023; Kuo et al., 2011; Lazzari et al., 2019; Liu et al., 2005; Yamashita et al., 2000; Yan et
73 al., 2006; Yoon et al., 2023). It is expected that the PAH record in coastal sediments cores
74 also indicates pre-industrial human activities such as biomass and charcoal burning especially
75 during ancient times.

76 Osaka Bay is located directly next to Osaka City, which is part of the Kyoto-Osaka-
77 Kobe metropolitan area, the second most urbanized area in Japan. Osaka City has experienced
78 industrialization and urbanization since the 1890s CE and has emitted a large amount of trace
79 metals during the post-World War II economic growth period in Japan (Hosono et al., 2010;
80 Nitzsche et al., 2022; Yasuhara and Yamazaki, 2005). Tsuji et al. (2020) found that PAH
81 concentrations in a young (< 50 years) sediment core from Osaka Bay peaked around 1980
82 CE and declined since then until now. Ishitake et al. (2007) and Moriwaki et al. (2005)
83 suggested that the burning of wooden structures followed by the intense bombing of Osaka
84 City led to a peak in PAHs around 1945 in a sediment core from Nagaike Pond and from
85 Osaka Castle moat, respectively. Moriwaki et al. (2005) also noted that increasing PAHs
86 concentrations in pre-industrial times (1680 to 1880 CE) were attributed to wood burning by
87 humans. Yet, such small catchments could be impacted by nearby events like fires.
88 Information on the historical trend of PAHs is present in the 17th century CE. Osaka,
89 formerly known as Naniwa, has a long history dating back to the Kofun period (250 to 592
90 CE), was scenery of repeated warfare during the Sengoku period (1467 to 1615 CE),
91 experienced an increase in the population and metal smelting activities and became an
92 important economic hub during the Edo period (1603 to 1867 CE) (Table 1). Thus, early

93 human activities such as wood and charcoal burning for domestic fires and smelting could
94 have led to high PAH emissions.

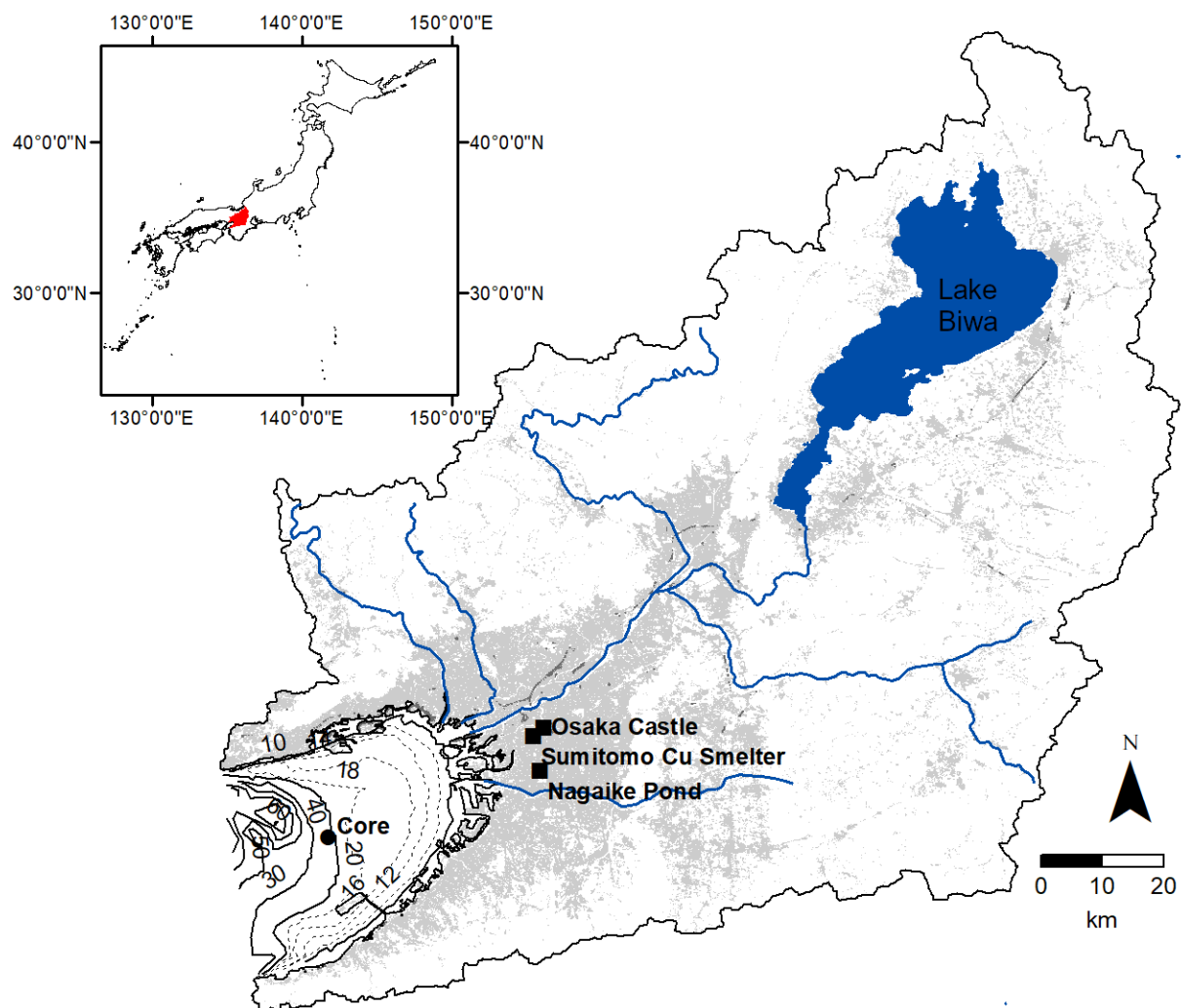
95 In this study, we investigated the temporal variation in PAH concentrations and
96 diagnostic ratios in a 9 m-long sediment core from Osaka Bay, which records a history of the
97 last 2,400 years. The aim of this study was to assess if the historical trend of PAHs can be
98 related to past human activities. In our previous study, we found increasing trace metal
99 concentrations in the core from the early 17th century CE due to early human activities
100 (Nitzsche et al., 2022). Thus, our second goal was to compare the PAH concentration to trace
101 metal concentrations in order to identify similar trends and sources.

102

103 **Study area and sampling**

104 *Osaka Bay and its catchment*

105 The catchment of Osaka Bay is located in Central Japan and covers a total area of
106 approximately 10,700 km² with the Yodo River catchment being the largest (8,240 km²).
107 Coniferous forests dominate the catchment of Osaka Bay; yet large urban and build-up land
108 areas are present owing to the Kyoto-Osaka-Kobe (Keihanshin) metropolitan area. Lake Biwa
109 is located in the northeastern part of the catchment with an area of approximately 3,170 km².
110 The geology of the whole catchment consists mainly of Quaternary unconsolidated sediments
111 (Itihara et al., 1988), clastic (meta)sedimentary rocks of the Jurassic and Cretaceous
112 accretionary complexes, Cretaceous granitoid rocks, felsic volcanic and volcanoclastic rocks,
113 and minor gneisses and schists, Carboniferous to Permian limestones and basalts, and
114 Cretaceous gabbro.



115
 116 Figure 1. Location of the Osaka Bay catchment in Japan (top left), and area of urban land in
 117 grey with major rivers and the locations of the sediment core, Osaka Castle, the Sumitomo Cu
 118 smelter and Nagaike Pond (bottom right). Note that Osaka Castle was constructed on the
 119 grounds of the former Ishiyama Hongan-ji temple. The former site of Naniwa-no-Miya Palace
 120 was directly adjacent to the later built Osaka Castle. Land-use data were obtained from Land
 121 Use Fragmented Mesh Version 2.5.1 from the National Land Numerical Information, created
 122 by Ministry of Land, Infrastructure, Transport and Tourism, the Government of Japan.

123
 124 *Sediment sampling*

125 A piston core was collected at station OS5B from the central part of Osaka Bay at a
 126 water depth of 24 m during cruise KT-11-13 on July 1, 2011 (Fig. 1). The core was cut in 2

127 cm-slices from the surface to the final core depth of 884 cm (see for further details Nitzsche et
128 al., 2022).

129

130 *Historical events in the catchment*

131 Since the early Holocene, about 7000 BCE, Osaka Bay and its catchment underwent
132 many natural (Kajiyama and Itihara, 1972; Matsuda, 2008; Yasuhara et al., 2002) and
133 artificial transformations (Pearson, 2016) as well as other human activities, which affected the
134 sedimentation rates and sediment sources, and probably the PAH input into the bay. In
135 particular, the brackish Kawachi Lagoon was transformed into Kawachi Lake during 400
136 BCE to around 1 BCE, into the marshland Kawachi Plain from around 800 CE, and finally
137 into the present Osaka plain from the second half of the 19th century CE (Kajiyama and
138 Itihara, 1972; Matsuda, 2008). Permanent habitation on the Uemachi Plateau and around the
139 Kawachi Plain started as early as 700 BCE (Pearson, 2016). The major events have been
140 reported in our previous work (Nitzsche et al., 2022), however, they are summarized in the
141 current study (Table 1), and the situation during the 3rd and 4th centuries CE is shown in
142 Figure S1.

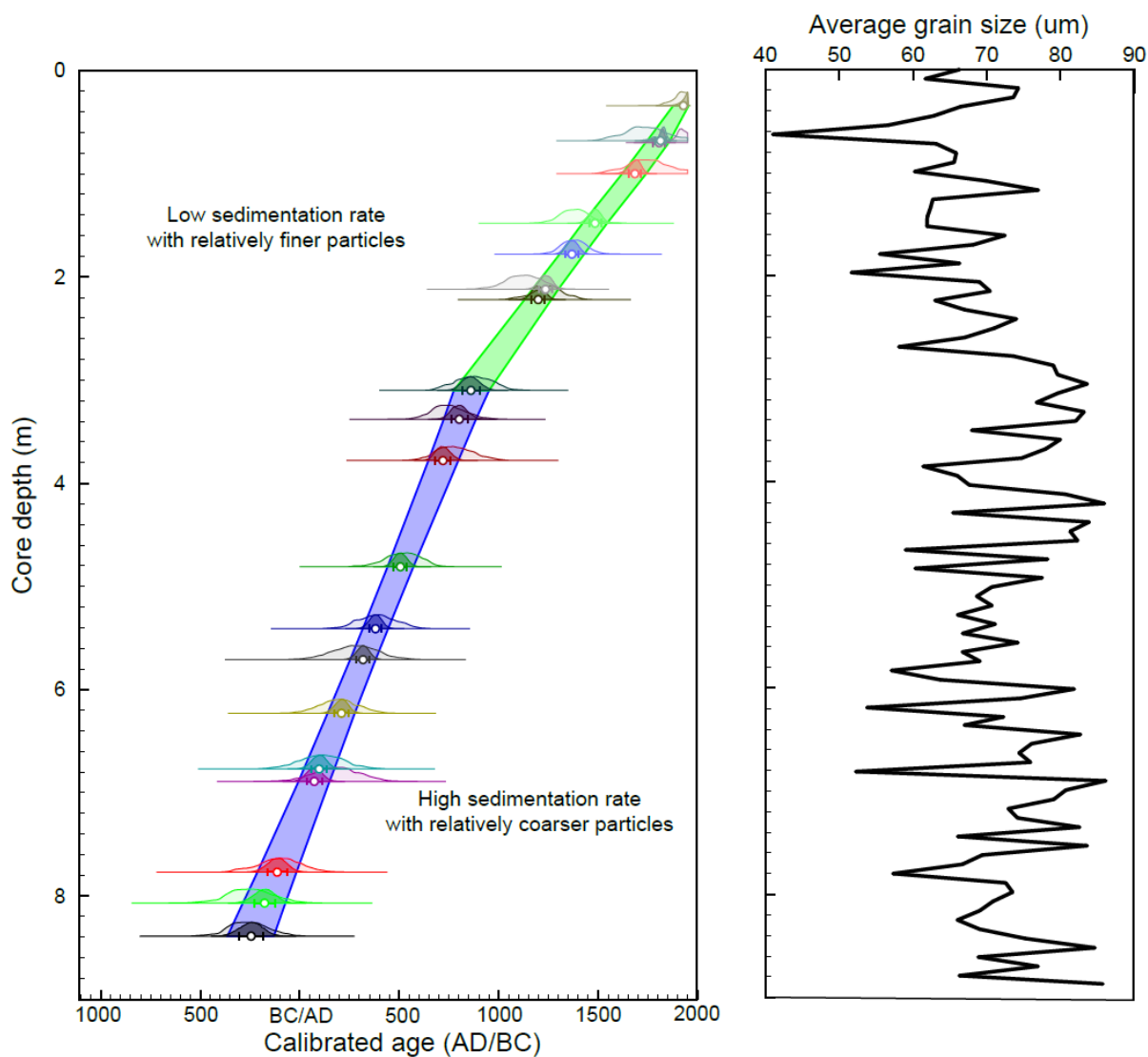
143 Table 1. Historical events in Osaka (Naniwa) City and the Osaka Bay catchment.

Period	Year	Event
Modern (1868 CE to present)	1960s CE	Transition from coal to oil
	1945 CE	Air raids destroyed large areas
	1940-1945 CE	WWII
	1929-1932 CE	Economic recession
	1914-1918 CE	WWI
	1904/1905 CE	Russo-Japanese War
	1894/1895 CE	First Sino-Japanese War
	1890s CE	Start of industrialization
	1868 CE	Opening of the port for trade
	Feudal (1185 to 1868 CE)	1863 CE
1837 CE		Big city fire
1724 CE		Big city fire
1690 CE		Expansion of the Sumitomo Copper Refinery
1620s CE		Reconstruction of Osaka Castle
1614/1615 CE		Siege of Osaka Castle
1583 CE		Construction of Osaka Castle
1569-1580 CE		Siege on Ishiyama Hongan-ji temple
1492 CE		Completion of Ishiyama Hongan-ji temple
Imperial and aristocracy (592 to 1185 CE)		1180 CE
	726 CE	Completion of Naniwa-no-Miya Palace
	686 CE	Naniwa Nagara-Toyosaki Palace burned down
	652 CE	Completion of Naniwa Nagara-Toyosaki Palace
	593 CE	Completion of Shitenno-ji temple
Transition (Kofun) (250 to 592 CE)	5 th century CE	Digging of the canal Naniwa no Horie
		Construction of the port facility Naniwa-tsu
		Start of iron smelting (<i>tatara</i>)
Yayoi (930 BCE to 250 CE)	From 700 BCE	Rice cultivation and permanent habitation

145 **Material and methods**

146 *Core dating*

147 The dating of the core has been previously described (Nitzsche et al., 2022). Briefly,
148 the piston core has been dated by 22 radiocarbon measurements of molluscan shells using
149 accelerator mass spectrometry (AMS) at the Micro Analysis Laboratory of the University of
150 Tokyo (Matsuzaki et al., 2004; Yokoyama et al., 2019). The technique employed for sample
151 preparation was previously reported (Yokoyama et al., 2007). The radiocarbon age results
152 were calibrated to calendar years using the OxCal ver. 4.4 software (Bronk Ramsey, 2009)
153 with the Marine 20 dataset (Heaton et al., 2020) assuming a regional specific reservoir (DR)
154 correction of 135 ± 20 years (Kuwae et al., 2013), which was estimated for Beppu Bay,
155 located in the Seto Inland Sea which is about 350 km west from our study site. The core
156 provided a continuous environmental record of the last 2,300 years (Fig. 2).



157

158 Figure 2. Core depth (cm) versus calibrated age of the piston core (a). For each ^{14}C age
 159 controlling point, P-sequence modeled and unmodeled age probability distributions are
 160 shown. The error bars represent the 1σ error of the P-sequence modeled age. Average grain
 161 size (b). Figure reproduced from Nitzsche et al. (2022).

162

163 *Core logging*

164 The sedimentary facies comprised of a homogenous olive grey mud inferred by X-ray
 165 computed tomography (CT) measurement at Kochi Core Center, JAMSTEC. The wet density
 166 of the core was determined via gamma ray transmission using a Multi-Sensor Core Logger
 167 (Geotek Ltd., Daventry, UK) at Kochi Core Center.

168

169 *Total organic carbon concentration and stable isotope analysis*

170 The total organic carbon (TOC) and carbon stable isotope ratios ($\delta^{13}\text{C}$) were
171 determined for 61 sediment samples from selected horizons after freeze-drying, sieving to <
172 125 μm , and manually powdering. TOC concentrations and $\delta^{13}\text{C}$ values were determined after
173 weighing samples into pre-cleaned smoothed wall tin capsules, decalcification with 0.5M HCl
174 and drying on a hotplate at 80 °C. The tin capsules with the dried samples were sealed, and
175 analyzed using a sensitivity-improved elemental analyzer (Flash EA1112, Thermo Finnigan,
176 Bremen, Germany) connected to an isotope ratio spectrometer (Delta plus XP, Thermo
177 Finnigan, Bremen, Germany) at the Biogeochemistry Research Center (BGC), JAMSTEC,
178 according to Ogawa et al. (2010). The isotopic values are expressed in delta notation (‰),
179 relative to VPDB (Vienna Pee Dee Belemnite) standard. The carbon isotopic compositions
180 were calibrated using five inter-laboratory determined standards, which ranged from -
181 26.86 ‰ to 0.18 ‰ (BG-T: L-tyrosine, BG-A: L-alanine, BG-P: L-proline, Tayasu et al.,
182 2011; L-Valine and L-Glutamic acid, Sun et al., 2023, Shoko Science Co., Ltd.). Analytical
183 errors of the $\delta^{13}\text{C}$ values estimated with repeated analyses of BG-T was ± 0.20 ‰ (1σ ; $n=14$).

184

185 *Extraction and analysis of polycyclic aromatic hydrocarbons*

186 Fifty-six samples were selected for PAH analysis to have one or two samples during
187 each century before the 17th century CE, and every 10 to 20 years after the 17th century CE.
188 The lipids of the samples (1.0 or 2.0 g) were extracted twice for 15 min by sonication using a
189 mixed solvent of dichloromethane/methanol (70:30, v/v). The extracts were evaporated, re-
190 dissolved in hexane, and loaded on deactivated 1 % H₂O silica gel columns pre-conditioned
191 with hexane. The N-1 fraction (hydrocarbons) was extracted with *n*-hexane, and the N-2
192 fraction (ketone, aldehyde, ester) with *n*-hexane/dichloromethane (50:50, v/v). Subsequently,
193 the N-1 and N-2 fractions were analyzed for PAHs using gas chromatograph-mass

194 spectrometry (GC-MS) (Agilent 7890A-GC with Agilent 5975C inert XL MSD) equipped
195 with the VF-5ms column (0.25 mm \times 30 m, film thickness 0.10 μ m) at the BGC. The oven
196 temperature was programmed as follows: maintained at 40 $^{\circ}$ C for 2 min, raised up to 120 $^{\circ}$ C
197 at 30 $^{\circ}$ C min^{-1} , then to 320 $^{\circ}$ C at 6 $^{\circ}$ C min^{-1} , and maintained at 320 $^{\circ}$ C for 20 min. Helium
198 was used as the carrier gas with a constant flow rate of 1 mL min^{-1} . A standard containing
199 naphthalene (Nap), acenaphthylene (Acy), acenaphthene (Ace), fluorene (Flu), phenanthrene
200 (Phe), anthracene (Ant), fluoranthene (Flt), pyrene (Pyr), benz[a]anthracene (BaA), chrysene
201 (Chr), benzo[b]fluoranthene (BbF), benzo[k]fluoranthene (BkF), benzo[a]pyrene (BaP),
202 indeno[1,2,3-cd]pyrene (IcdP), dibenz[a,h]anthracene (DahA) and benzo[g,h,i]perylene
203 (BghiP)) (AccuStandard, USA) was prepared and analyzed together with the samples for
204 identification and quantification of PAHs (Table 2). Additionally, a perylene standard was
205 analyzed. Peak areas of individual PAHs were obtained in selected ion monitoring (SIM)
206 mode and used for the quantification. The concentration of PAHs in three analytical blanks
207 were below the detection limit. To account for any losses of individual PAHs during the
208 sample preparation, a PAH standard was evaporated and treated as the samples. The recovery
209 rates ranged from 70 to 90 % for 3-ring PAHs, from 90 to 100 % for 4-ring PAHs, and were
210 around 100 % for 5- and 6-ring PAHs.

211 We calculated the ratio of Flt/(Flt+Pyr) as a proxy for PAH sources, i.e., a ratio
212 Flt/(Flt+Pyr) smaller than 0.4 indicates petroleum leakage, between 0.4 to 0.5 is indicative of
213 petroleum combustion, and larger 0.5 is characteristic of biomass and coal combustion
214 (Yunker et al., 2002). A IcdP/(IcdP+BghiP) ratio of below 0.2 is indicative of petroleum,
215 between 0.2 and 0.5 of petroleum combustion, and greater than 0.5 of biomass and coal
216 combustion (Yunker et al., 2002). Furthermore, we calculated the ratio of the low molecular
217 weight PAHs Phe + Ant + Flt + Pyr (LMW) relative to the sum of PAHs (LMW/total).
218 Smoke-derived PAHs have a LMW/total ratio between 0.75 and 0.9, and combustion residues
219 have between 0.3 and 0.8, respectively (Karp et al., 2020). We also compared the total

220 concentration of PAH (Σ PAH) with metal concentrations in the same sediment core that
 221 were reported in our previous paper (Nitzsche et al., 2022) to estimate PAH sources.

222
 223 Table 2. List of the 17 polycyclic aromatic hydrocarbons (PAH) analyzed in this study, with
 224 their number of rings and molecular weight.

Compound	Abbreviation	Rings	Molecular weight (g mol ⁻¹)
Naphthalene	Nap	2	128
Acenaphthylene	Acy	3	152
Acenaphthene	Ace	3	154
Fluorene	Flu	3	166
Phenanthrene	Phe	3	178
Anthracene	Ant	3	178
Fluoranthene	Flt	4	202
Pyrene	Pyr	4	202
Benz[a]anthracene	BaA	4	228
Chrysene	Chr	4	228
Benzo[b]fluoranthene	BbF	5	252
Benzo[k]fluoranthene	BkF	5	252
Benzo[a]pyrene	BaP	5	252
Perylene*	Per	5	252
Benzo[g,h,i]perylene	BghiP	6	278
Dibenz[a,h]anthracene	DahA	5	278
Indeno[1,2,3-cd]pyrene	IcdP	6	276

225 * Not included in the sum of total PAHs

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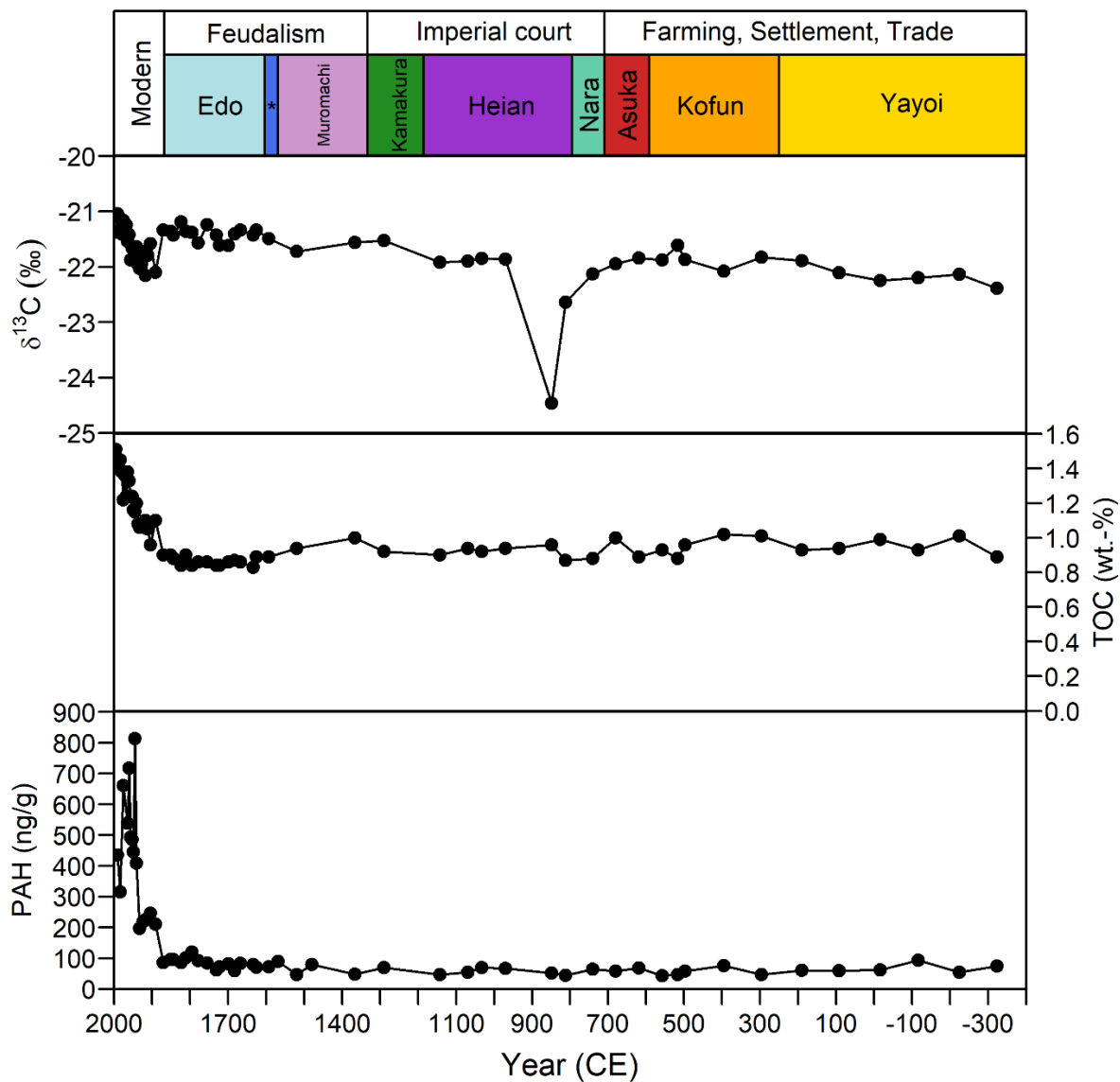
227 Results

228 *Density, carbon stable isotopes and total organic carbon*

229 The wet density was relatively constant and usually ranged from 1.3 to 1.7 g cm⁻³ (Fig.
 230 S2). The $\delta^{13}\text{C}$ values varied from -22.7 ‰ to -21.0 ‰ except for an outlier (-24.5 ‰) around
 231 the beginning of the Heian period at approximately 850 CE (Fig. 3). The TOC content varied
 232 between 0.83 % and 1.02 % from around 320 BCE to 1870 CE. Since then, the TOC content
 233 continuously increased to 1.51 % at the surface.

234 We found that the only little variations in the wet density, the $\delta^{13}\text{C}$ values and TOC %,
 235 the homogenous olive grey mud, the relatively constant grain size distribution, and the

236 constant sediment accretion rate (SAR) except for a drop around 800 CE (Fig. 2) (Nitzsche et
 237 al., 2022), suggesting that the grain density and porosity do not drastically change throughout
 238 the core. Therefore, we suggest an almost constant mass accumulation rate (MAR) and that
 239 the PAH concentrations are roughly parallel with the MAR.



240

241 Figure 3. Temporal trend of $\delta^{13}\text{C}$ values, total organic carbon (TOC), and the sum of
 242 polycyclic aromatic hydrocarbons (PAH) against total dry sample. *Azuchi-Momoyama
 243 period.

244

245 *Polycyclic aromatic hydrocarbons*

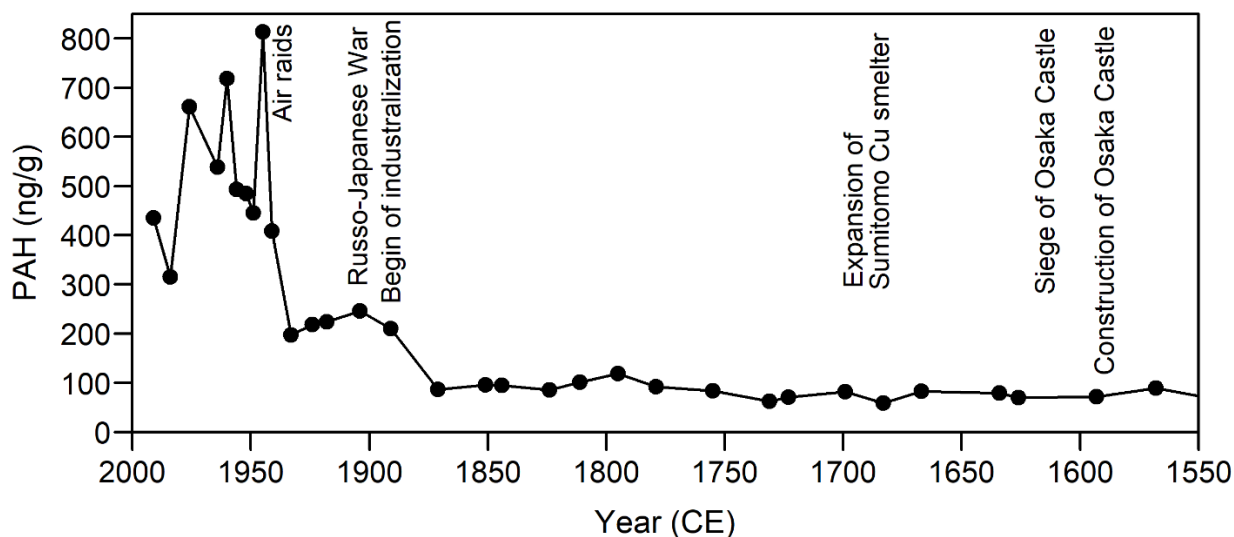
246 The low-weight three-ring PAHs naphthalene, acenaphthylene, acenaphthene, and
247 fluorene were often below the detection limit. Thus, the sum of PAHs (Σ PAH) in this study
248 involved the following 12 PAHs: phenanthrene (Phe), anthracene (Ant), fluoranthene (Flt),
249 pyrene (Pyr), benz[a]anthracene (BaA), chrysene (Chr), benzo[b]fluoranthene (BbF),
250 benzo[k]fluoranthene (BkF), benzo[a]pyrene (BaP), benzo[g,h,i]perylene (BghiP),
251 dibenz[a,h]anthracene (DahA), and indeno[1,2,3-cd]pyrene (IcdP). Overall, the Σ PAH
252 concentrations ranged from about 40 to 800 ng g⁻¹ throughout the core (Fig. 3). From the
253 Yayoi to the end of the Azuchi-Momoyama period, between 320 BCE and 1603 CE, the Σ
254 PAH concentration ranged between 43 and 91 ng g⁻¹ without a clear pattern. The Σ PAH
255 concentration started to increase from the early 1700s CE to about 1800 CE. Since then, it
256 remained constant until about 1870 CE (Fig. 4). From then, the Σ PAH concentration
257 increased until the early 1900s CE and remained constant again until the early 1930 CE. The
258 peak in the Σ PAH concentration was observed at around 1945 CE and started to scatter
259 towards the surface.

260 With respect to individual PAHs, Phe showed a relatively large scatter throughout the
261 core (Fig. S3). Anthracene, Flt, and Pyr tended to increase towards the surface from the 1870s
262 CE, and BaA and Chr increased from the early 18th century CE. Higher mass PAHs (> 228 g
263 mol⁻¹) started to increase from the early 17th century CE.

264 The ratio of Flt/(Flt+Pyr) was relatively stable throughout the core (~0.60) until the
265 end of the 17th century. Since then, a slight decrease to ~0.5 until the 20th century was
266 observed (Fig. 5). Similarly, the ratio of LMW/total was relatively constant (~0.75) until the
267 beginning of the 17th century CE and decreased to ~0.4 until the top of the core (Fig. 5). The

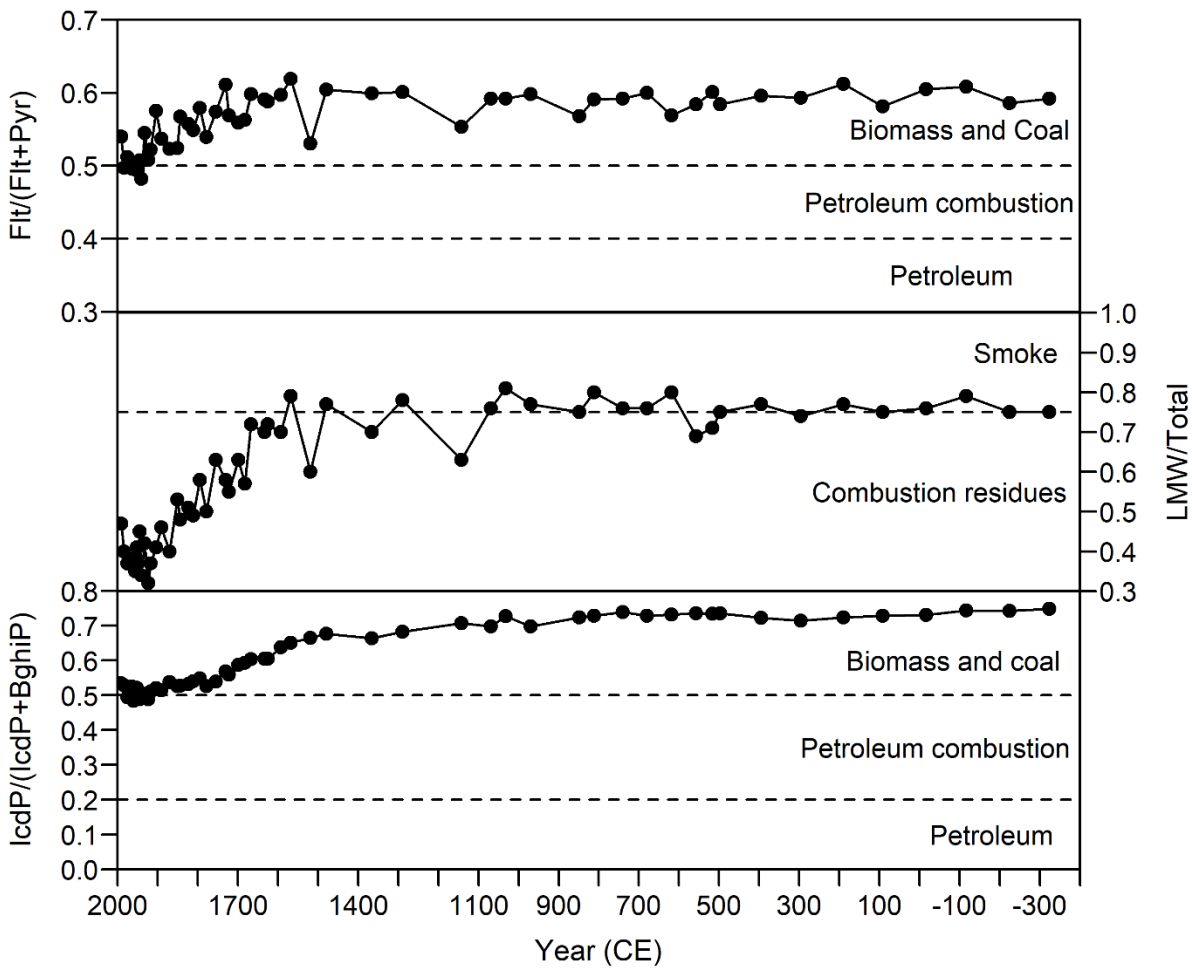
268 IcdP/(IcdP+BghiP) ratio was around 0.71 until the beginning of the 17th century CE and
 269 decreased to approximately 0.5 towards the surface.

270 We found the logarithm of the \sum PAH concentration was positively correlated with the
 271 logarithm of the Zn, Cu, and Pb concentrations (Fig. 6) determined in our previous study
 272 (Nitzsche et al., 2022).



273
 274 Figure 4. Temporal trend of the sum of polycyclic aromatic hydrocarbons (PAH) from 1550
 275 to 2000 CE with major historical events in the catchment or Japan.

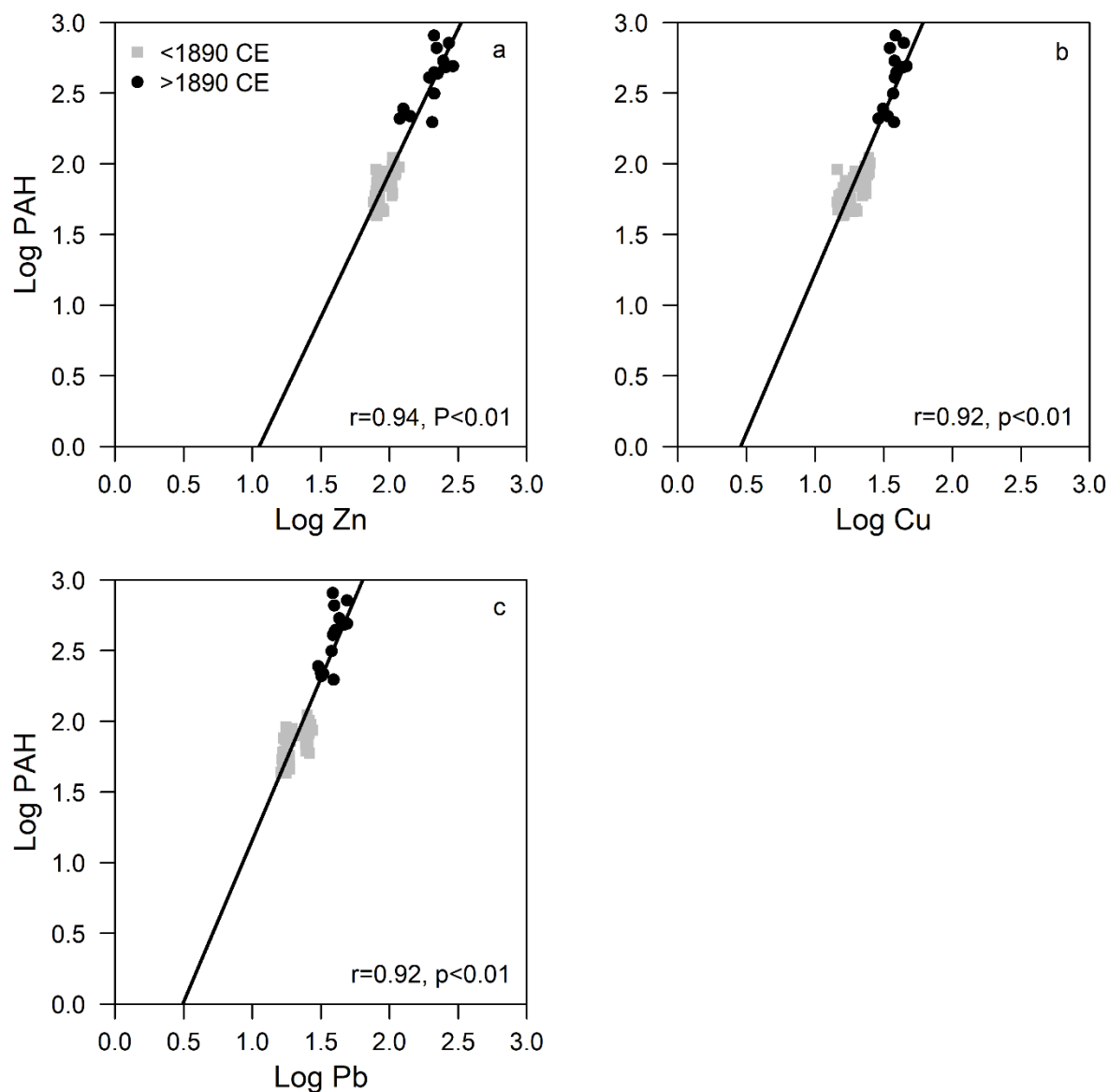
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277

278 Figure 5. Historical trend of the ratios $Flt/(Flt+Pyr)$, $LMW/total$, and $IcdP/(IcdP+BghiP)$.

279



280

281 Figure 6. Scatterplot of the logarithm of the sum of polycyclic aromatic hydrocarbons (PAHs)

282 against the logarithm of zinc (a), against the logarithm of copper (b), and against the

283 logarithm of lead (c). The original Zn, Cu and Pb concentrations are reported in Nitzsche et

284 al. (2022).

285

286 **Discussion**287 *Historical trend of PAHs before the Edo period*288 Based on the temporal changes in the Σ PAH concentration and composition, we gain

289 insights into human activities in the catchment of Osaka Bay throughout time. Nevertheless,

290 the low Σ PAH concentration of < 100 ng and the lack of pattern before the 17th century CE
291 (Fig. 3) could be partly attributed to diagenetic alteration (Wakeham et al., 1980). For
292 instance, the diagnostic ratios of Ant/(Ant+Phe) and BaA/(BaA+Chr) (Yunker et al., 2002),
293 indicate petroleum before the 19th CE (Fig. S4). Yan et al. (2005) attributed similar unrealistic
294 patterns of Ant/(Ant+Phe) and BaA/(BaA+Chr) found in a 200-year-old sediment core from
295 Central Park Lake, New York City, to the higher photolytic breakdown of Ant and BaA
296 relative to Phe and Chr. Nevertheless, Ant and BaA contribute less than 10 % to the Σ PAH
297 concentration. Furthermore, small changes in the MAR could have caused dilution effects,
298 which could not be seen due to the temporal resolution of the core. The pattern of the Σ PAH
299 concentration from the beginning of the 17th century CE agreed to historical events.
300 Therefore, we consider diagenetic alteration to be minimal across the core. The increasing
301 concentration of perylene with increasing core depth (Fig. S5) agrees with previous studies,
302 which suggest a contribution of in-situ production of perylene by microbial communities
303 under anoxic conditions to the sediments (Gschwend and Hites, 1981; Silliman et al., 2001;
304 Slater et al., 2013). Furthermore, some perylene could have been derived from terrestrial
305 fungi (Grice et al., 2009; Itoh and Hanari, 2010).

306 The LMW/total ratios around 0.75 indicate that PAHs were derived from smoke rather
307 than combustion residues before the Edo period (Fig. 5) (Karp et al., 2020). Such PAHs likely
308 represent the deposition of natural PAHs derived from wildfires as well as anthropogenic
309 PAHs derived from softwood and charcoal combustion. Recent work has tested the source
310 area (up to 150 km) of PAHs originating from wildfires (Vachula et al., 2020). Yet, it is
311 difficult for the current dataset to estimate the exact source area because PAHs could have
312 additional origins than the Osaka Bay catchment owing to their potential for long-range
313 atmospheric transport (Halsall et al., 2001). For instance, PAHs could have originated from
314 wildfires in other areas of Japan, in China, the Korean peninsula and even Eastern Russia,

315 which are common sources of PAHs in western Japan today (Inomata et al., 2012, 2013; Tang
316 et al., 2002; Yang et al., 2007). Thus, the scatter in the Σ PAH concentration during the pre-
317 Edo times is difficult to explain. It should be noted that we did not find any change in the
318 trace metal concentrations before the Edo period (Nitzsche et al., 2022).

319 Interestingly, the warring activities during the Azuchi-Momoyama period (1568 to
320 1600 CE), such as the siege of the Ishiyama Hongan-ji temple (1569 to 1580 CE), were not
321 reflected in the Σ PAH concentration. While these events likely emitted PAHs to the
322 atmosphere via the combustion of wood and charcoal for weapon and armor smelting and the
323 burning of military and civil wooden buildings, tracing their impacts on the sedimentary Σ
324 PAH concentration would require a higher temporal resolution and/or the PAH emission
325 might be too low to be detected.

326

327 *Historical trend of PAHs during the Edo period*

328 Human impacts on the PAH concentration and composition are more apparent during
329 the Edo period. Since the Osaka Castle was destroyed by the Tokugawa forces in 1615 CE,
330 the local population and economy have continuously grown during the peaceful Edo period
331 which made Osaka City an important center for trade, culture, and food products. The
332 increasing Σ PAH concentration from the early 18th century CE until approximately 1800 CE
333 (Fig. 4) agree with the increasing population in the Osaka area from the mid-17th century CE
334 until the end of the 18th century CE (Saito, 2002). The LMW/total ratio decrease from the
335 beginning of the 17th century CE (Fig. 5) indicates a shift from smoke to particulate PAHs.
336 Combustion residues from biomass and charcoal could have directly entered the rivers and
337 seawater and finally deposited in the sediment. During the Edo period, charcoal gained a
338 widespread importance, and a new type of white charcoal, the so-called *binchotan*, was

339 combusted after being transported from Wakayama Prefecture (Totman and Kumazaki, 1998).
340 Another PAH source could be the start of salt production in the Seto Inland Sea area, which
341 required the burning of large amounts of wood and charcoal leading to barren areas in the
342 mountains.

343 The increasing Σ PAH concentration from the early 18th century CE until
344 approximately 1800 CE is consistent with an increase in the Cu concentration observed in the
345 same core until the beginning of the 19th century CE (Nitzsche et al., 2022). This was
346 attributed to emissions from ore smelting by the Sumitomo Copper Refinery, established 1 km
347 southwest of Osaka Castle in 1655 CE and expanded in 1690 CE (Suzuki et al., 1998). Thus,
348 the burning of wood and charcoal for heat and energy generation and the early usage of oil
349 lamps are likely sources of human-derived PAHs. The constant Σ PAH concentration until
350 the late 19th century overlaps with a decline in population in the Osaka area, which started at
351 the end of the 18th century CE (Saito, 2002). The big city fires in 1724, 1837, and 1863 CE
352 could not be detected in the core which requires a higher temporal resolution and/or the PAH
353 emissions were too low to be detected.

354

355 *Historical trend of PAHs during the Modern age*

356 The rapid tripling of the Σ PAH concentration from the end of the 19th century CE
357 until the beginning of the 20th century CE is associated with increasing Cu, Zn, and Pb
358 concentrations in the core and marked the beginning of industrialization (Nitzsche et al.,
359 2022). The further decreasing LMW/total ratio from the 1870s CE could be due to the
360 combustion of imported and mined coal. The development of the iron and steel industry, Cu
361 smelters, shipbuilding, and other industries, as well as steam ships, emitted large amounts of
362 smoke into the atmosphere and combustion residues into the rivers and the bay. Furthermore,
363 Japan participated in the Shino-Japanese War (1894-1895), the Russo-Japanese War

364 (1904/1905) and World War I (1914 to 1918 CE). An increase in the Σ PAH concentration
365 was also observed in a core from Osaka Castle moat, yet about 25 years earlier (Moriwaki et
366 al., 2005), which could be due to uncertainties in age models, differences in the time of
367 deposition as well as local impacts in the vicinity of Osaka Castle. The relatively constant Σ
368 PAH concentration from the mid-1910s CE until the early 1930s CE could partly be explained
369 by a relatively stagnant economy in Japan (1929 to 1932 CE). A relatively constant Σ PAH
370 concentration was also observed between the early 1900s and the early 1920s CE in a core
371 from Tokyo Bay (Yamashita et al., 2000). The peak in the Σ PAH concentration around 1945
372 CE was also observed in the Osaka Castle moat (Moriwaki et al., 2005) and Nagaike Pond, a
373 reservoir in the South of Osaka City (Ishitake et al., 2007), which was explained by burning
374 of wooden buildings due to the heavy bombing raids on Osaka and Kobe City during World
375 War II. In addition, traffic-related PAHs became an important source after World War II. A
376 high Σ PAH concentration during the late 1970s CE was also detected in a coastal core close
377 to the Yodo River mouth and marked the beginning of the use of cleaner energy and coal
378 filters during the 1980s (Tsuji et al., 2020). The diagnostic Flt/(Flt+Pyr) and
379 IcdP/(IcdP+BghiP) ratios suggest biomass, coal, and petroleum combustion during the 20th
380 century CE (Fig. 5), which agrees with surface sediments in Osaka Bay (Tsuji et al., 2020).
381 Thus, the transition from coal to oil around 1950 CE is not clearly observed in the core.

382 Contemporaneous changes in the Σ PAH and trace metal, in particular Pb,
383 concentrations in sediment cores are often observed and explained by similar elemental
384 sources such as smelting (Azoury et al., 2013; Leorri et al., 2014; Musa Bandowe et al., 2014;
385 Wang et al., 2018). In the case of Osaka Bay, the contemporaneous increase in Σ PAH and
386 Cu, Zn, Pb concentrations from the late 19th century CE (Nitzsche et al., 2022) and peak

387 around 1960 CE is confirmed by positive correlations between the Σ PAH with Cu, Zn and
388 Pb concentrations in the sediment core (Fig. 6). The results indicate the combustion of wood
389 and charcoal during the 18th and 19th centuries and of coal from the late 19th century CE for
390 smelting and other industrial activities. The introduction of ash filters and transition from coal
391 to oil could have led to a decline in Σ PAH and trace metal concentrations from the early
392 1960s CE. In the case of trace metals, the decline was also promoted by the regulation of
393 sewage waters and industrial effluents. More insights into the shared PAH and Pb sources can
394 be gained by Pb stable isotopes (Leorri et al., 2014).

395

396 **Summary and conclusions**

397 This study successfully compared the PAH record in a coastal marine sediment core
398 with historical events and human activities. The total PAH concentration and LMW/total ratio
399 before the beginning of the 17th century CE were relatively constant and mainly reflected
400 smoke-derived PAHs from the natural background. A shift to a higher relative abundance of
401 concentrations of 4–6 ring PAHs from the early 17th century CE and a higher PAH
402 concentration from the early 18th century CE until approximately 1800 CE agreed with a
403 population increase and increasing importance of charcoal during the peaceful Edo period.
404 The constant PAH concentration until the late 19th century CE overlapped with a decline in
405 the local population, while the increasing PAH concentration from the late 19th century CE
406 marks the beginning of industrialization. The peak in PAH concentration in 1945 CE was
407 caused by burning of the wooden structures due to air raids on Osaka City.

408 The combination of PAHs with trace metals, in particular Cu and Pb, yielded insights into
409 related sources, i.e., combustion of wood and charcoal (PAH source) for smelting Cu ores (Cu
410 source) during the 18th century CE, and combustion of coal for smelting and heavy industries
411 from the end of the 19th century CE. Our results demonstrate that PAH concentrations and

412 diagnostic ratios in coastal sediment cores can be used to reconstruct pre-industrial human
413 activities.

414

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622

623 **Supplementary information**