Batteries & Supercaps

Supporting Information

SiCN Ceramics as Electrode Materials for Sodium/Sodium Ion Cells – Insights from ²³Na In-Situ Solid-State NMR

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Figure S1: a) Galvanostatic charge/discharge curves of the PDC material during the 1st, 5th and 50th cycle obtained at a mass specific current of 37.2 mA g⁻¹. b) Cyclic voltammogram of the PDC material obtained in the voltage range between 1 to -0.05 V vs. Na/Na⁺ with a rate of 20 μ V/s.



Figure S2: ²³Na *in-situ* NMR spectra for the Na|NaPF₆|SiCN electrochemical cell. The electrochemical cycling was conducted in the voltage window of -0.03 to 2.5 V applying a current of \pm 75 µA for 16.56 h. A 3D enlarged view of the metal and electrolyte region is presented in the main manuscript in Figure 1.



Figure S3: ²³Na *in-situ* NMR spectra of the Na|NaPF₆|SiCN cell obtained when galvanostatic cycling was performed with a current of $\pm 20 \mu$ A at a rate of C/30 corresponding to the graphite specific capacity of 372 mAhg⁻¹.



Figure S4: Enlarged 3D view of the ²³Na *in-situ* NMR spectra presented in Figure S3 for a) the electrolyte region (0 to -30 ppm) and b) the metal region (1150 to 1090 ppm). An increase respectively decrease of the shoulder signal at 1128 ppm during the desodiation and sodiation is visible.



Figure S5: ²³Na *in-situ* NMR spectra of the Na|NaPF₆|SiCN cell obtained when galvanostatic performance was carried out with a current of ±20 μ A at a rate of C/30 corresponding to the graphite specific capacity of 372 mAhg⁻¹. When the voltage minimum has been achieved, the cell was held at -0.03 V for around 10 hrs.



Figure S6: Enlarged 3D view of the ²³Na *in-situ* NMR spectra presented in Figure S5 for a) the electrolyte region (0 to -30 ppm) and b) the metal region (1150 to 1090 ppm). Along this performance, no significant changes were found in the ²³Na *in-situ* NMR spectra.



Figure S7: a) ²³Na *ex-situ* MAS NMR spectrum of the cycled SiCN material. The spectrum was recorded at 7 T at a spinning rate of 50 kHz with an offset of 200 ppm employing the Hahn-echo sequence. Note: Spinning sidebands are marked with asterisks. b) Zoom in of spectrum (a) (red spectrum) and of the spectrum of pristine SiCN material wetted with the electrolyte solution (black spectrum). Note: Both spectra were measured with the same acquisition parameters and are normalized to their maximum height. c) Comparison of the spectra of the cycled SiCN material recorded at 7 T at a spinning rate of 50 kHz with single pulse excitation employing short flip angles of 0.1 and 0.2 μ s (blue and red spectra) and 81920 respectively 20480 scans. Note that the spectrum measured with a flip angle of 0.1 μ s is scaled by a factor of 1.65 to ease the comparison.



Figure S8: a) An example for the performed deconvolution of the ²³Na *in-situ* NMR spectra by three Gaussian/Lorentzian lines. The experimental spectrum is illustrated as red curve and the cumulative fit peak is displayed as blue curve. Enlarged view of the ²³Na *in-situ* NMR spectrum for b) the electrolyte region (0 to -25 ppm), and c) the metal region including the shoulder structure (1150 to 1105 ppm).

Parameter	Chemical shift (ppm)	FWHM (ppm)	Rel. Area (%)
Fit 1	-13.55	4.00	78.83
Fit 2	1120.84	3.82	17.25
Fit 3	1127.68	13.96	3.92

Table S1: Fit data for the ²³Na *in-situ* NMR spectrum shown in Figure S7.



Figure S9: ¹⁹F *ex-situ* MAS NMR spectra of a) the wetted SiCN with NaPF₆ and b) the cycled SiCN material. Spectra were recorded at 14.1 T at a spinning frequency of 20 kHz.



Figure S10: Enlarged view of the ¹⁹F *ex-situ* MAS NMR spectra of a) NaF and b) cycled SiCN shown in S9.

The ¹⁹F *ex-situ* MAS NMR spectra were acquired on a 600 MHz Bruker Avance III HD spectrometer at 14.1 T corresponding to a frequency of 564.68 MHz for ¹⁹F. The measurements were performed using a 1.3 mm MAS rotor at a spinning frequency of 20 kHz and applying single pulse excitation. For the recording of the ¹⁹F spectra a pulse length of 2.1 μ s and a recycle delay of 3 s were set accumulating 80 scans. The ¹⁹F chemical shift was referenced relative to CFCl₃ using the signal of the polytetrafluorethylene (PTFE) at -122.7 ppm as external reference.