



Supporting Information

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Rings and Chains: Synthesis and Characterization of
Polyferrocenylmethylene

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Volker Presser, Markus Gallei,* and André Schäfer*

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X-ray diffraction data

CCDC code	2049104
Empirical formula	$C_{93}H_{120}Fe_6$
Formula weight	1572.98
Temperature	152(2) K
Wavelength	0.71073 Å
Crystal system	trigonal
Space group	<i>R</i> -3c
Unit cell dimensions	$a = 22.7410(10)$ Å $\alpha = 90^\circ$ $b = 22.7410(10)$ Å $\beta = 90^\circ$ $c = 12.4656(6)$ Å $\gamma = 120^\circ$
Volume	$5582.9(6)$ Å ³
Z	3
Density (calculated)	1.404 mg m ⁻³
Absorption coefficient	1.187 mm ⁻¹
F(000)	2502
Crystal size	$0.552 \times 0.405 \times 0.220$ mm ³
Theta range for data collection	1.791 to 36.163°
Index ranges	$-35 \leq h \leq 37$, $-37 \leq k \leq 37$, $-20 \leq l \leq 20$
Reflections collected	58994
Independent reflections	5931 [R(int) = 0.0240]
Completeness to theta = 25.242°	100.0%
Absorption correction	semi-empirical from equivalents
Max. and min. transmission	0.7471 and 0.6826

Refinement method	full-matrix least-squares on F^2
Data / restraints / parameters	5931 / 13 / 203
Goodness-of-fit on F^2	1.070
Final R indices [$I > 2\sigma(I)$]	R1 = 0.0321, wR2 = 0.0959
R indices (all data)	R1 = 0.0380, wR2 = 0.1006
Extinction coefficient	n/a
Largest diff. peak and hole	0.793 and -0.834 e.Å ⁻³

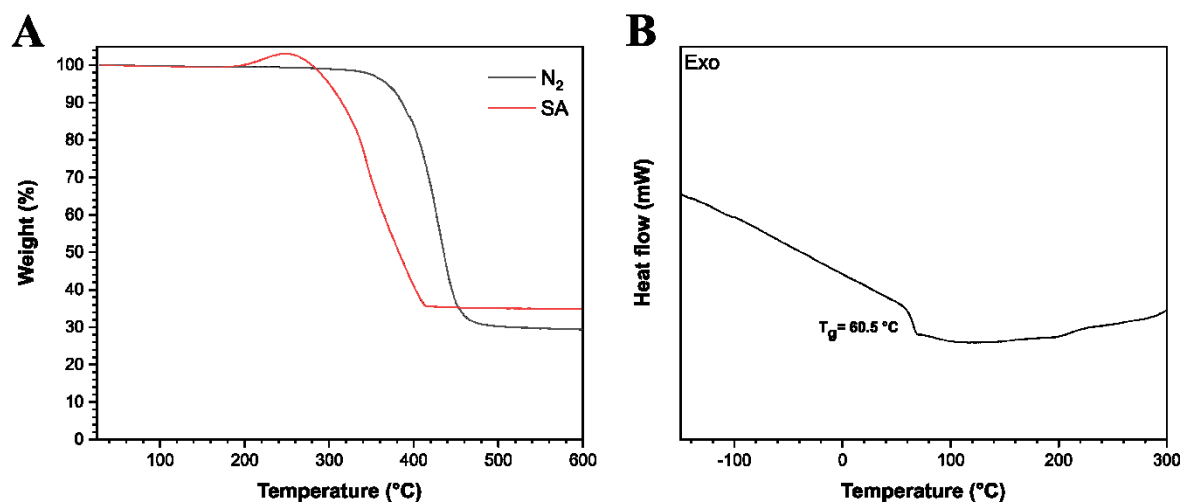


Figure S1. A: TGA curves of polyferrocenylmethylene (PFM) in an atmosphere of nitrogen (N₂, black curve) and synthetic air (SA, red curve) using a heating rate of 20 K min⁻¹, starting from 30 °C to 600 °C and B: DSC thermogram of PFM.

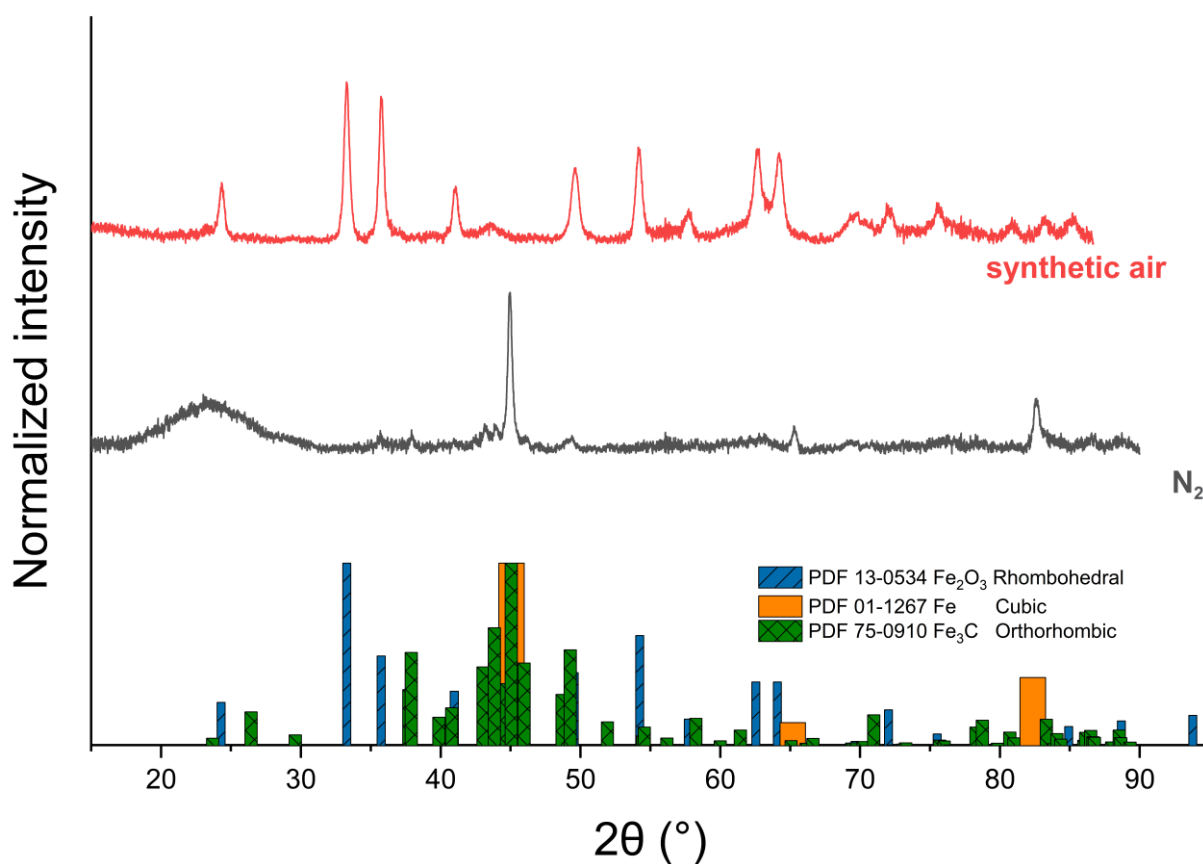


Figure S2. X-ray diffraction pattern of the residues after thermogravimetric analysis and corresponding reference data.